## THE TOTAL SYNTHESIS OF (±)-CORIOLIN\*

Sir:

Coriolin (1), a metabolite of the Basidiomycete, *Coriolus consors*, was isolated by H. UMEZAWA and coworkers in 1969<sup>1)</sup> and its structure was established in 1971 through chemical studies,<sup>2)</sup> followed by X-ray crystallographic studies in 1974<sup>3)</sup>. The unusual antitumor activity and the unique tricyclo 5-5-5- fused ring system have attracted widespread interest. We have already reported a stereocontrolled total synthesis of hirsutene (2)<sup>4)</sup>, a biogenetic precursor of coriolin<sup>5)</sup>, by using a unique skeletal-rearrangement of a tricyclo 6-4-5- fused ring derivative (3) to a tricyclo 5-5-5- fused ring derivative (4).

We now wish to record the first total synthesis of  $(\pm)$ -coriolin (1).

The key epoxide 4, upon treatment with NaI and zinc in 97% aq.DMF (130°C, 2 days), afforded the olefin  $5^{4}$ , which was hydrolized with 1% sulfuric acid in aq.acetone (50°C, 1 day) to the

ketoalcohol 6 (72% overall yield from 4); mp 82.5~83.5°C. A catalytic OsO4 cis-dihydroxylation<sup>6)</sup> of 6 (0.1 eq. OsO<sub>4</sub>, 8.5 eq. N-methylmorpholine-N-oxide, room temp., 40 hours) to give the triol 7; mp 108~109.5°C, PMR (acetone-d<sub>6</sub>)  $\delta$  1.01 (3H, s, Me), 1.04 (6H, s, Me  $\times$  2), 3.07 (1H, d,  $J_{2,6} = 9$  Hz, H-2), 3.95 (1H, dd,  $J_{6,7} = 5.5 \text{ Hz}$ ,  $J_{7,0H} = 5.0 \text{ Hz}$ , H-7), 4.26 (1H, d, OH-7), followed by treatment with 2,2dimethoxypropane and TsOH (room temp., 10 minutes) gave the acetonide 8 as cubic crystals; mp 164.5~166°C (85% overall yield). The stereochemistry of the newly introduced hydroxyl groups at C-7 and C-8 positions was controlled by that of the C-1-methyl group because of the difficulty in forming a trans 5-5 - fused ring<sup>7)</sup> and consequently fixed in cis configurations relative to the angular methyl group. Compound 8 was ox-

\* Dedicated to Professor SUMIO UMEZAWA on his 70th birthday.

idized with pyridinium chlorochromate (room temp., 1 hour) to give the ketone **9** in a nearly quantitative yield; mp 163.5~165°C, PMR (CDCl<sub>3</sub>)  $\delta$  1.09 and 1.16 (each 3H, s, Me × 2), 1.38 (9H, s, Me and acetonide), 3.00 (1H, d, J<sub>2,6</sub>=9.0 Hz, H-2), 4.38 (1H, d, J<sub>6,7</sub> = 5.0 Hz, H-7); which was the key intermediate for the following functionalization. Bis-sulfenylation<sup>8)</sup> of **9** with NaH (2.4 eq.) and methyl 2-nitrophenyl disulfide (mp 58~59°C, 2.4 eq.) in THF (room temp., 13 minutes) gave the thioketal **10**; mp 162.5~163.5°C, pmr (CD-Cl<sub>3</sub>)  $\delta$  1.09, 1.15 and 1.50 (each 3H, s, Me × 3), 1.34 (6H, s, acetonide), 2.14 (6H, s, SMe × 2),

Chart 1.





2.65 (2H, AB-q, CH<sub>2</sub>-9), 4.43 (1H, d, J<sub>6,7</sub>=5 Hz, H-7); which was treated with thallium (III) trinitrate (2.2 eq.) in MeOH (room temp., 5 minutes) to give the corresponding dimethylketal 11 (54% overall yield); mp 131.5~132.5°C, **PMR** (CDCl<sub>3</sub>)  $\delta$  1.08, 1.15 and 1.44 (each 3H, s, Me × 3), 1.35 (6H, s, acetonide), 2.96 (1H, d,  $J_{2,6} = 9$  Hz, H-2), 3.25 and 3.37 (each 3H, s, OMe  $\times$  2), 4.40 (1H, d, J<sub>6,7</sub> = 5 Hz, H-7). The latter was transformed by treatment with excess methyl lithium in ether  $(-70^{\circ}C \text{ to } 0^{\circ}C, 1 \text{ hour})$  to the tertiary alcohol 12 (60% yield); mp 186~187°C, PMR (CDCl<sub>3</sub>)  $\delta$  1.04, 1.05, 1.10, 1.19, 1.32 and 1.41 (each 3H, s, Me  $\times$  4 and acetonide), 2.0 (2H, m, CH<sub>2</sub>-5), 2.29 (2H, AB-q, J=15 Hz, CH<sub>2</sub>-9), 3.05 (1H, m, H-6), 3.34 (6H, s, OMe  $\times$  2), 3.56  $(1H, d, J_{2,6} = 10.5 \text{ Hz}, H-2), 4.40 (1H, d, J_{6,7} =$ 6 Hz, H-7); which was reduced in a small amount of THF by excess Li and liquid ammonia (-30)°C, 5 minutes) to give the desired  $\alpha$ -alcohol 13 (78% yield); mp 183.5~185.5°C, PMR (CDCl<sub>3</sub>)  $\delta$  0.92, 1.04, 1.17, 1.20, 1.32 and 1.49 (each 3H, s, Me × 4 and acetonide), 2.55 (1H, m, H-2), 3.34 (6H, s, OMe  $\times 2$ ), 3.68 (1H, dull dd,  $J_{2,3} = 9.5$ Hz,  $J_{3,OH} = \sim 3$  Hz, H-3), 4.14 (1H, d,  $J_{6,7} = 6$ Hz, H-7). Conclusive evidence for the stereochemical assignments at the C-3 and C-11 positions was provided by transformation of 13 to the unsaturated ketone 16 which was identical with the naturally derived product as described later on. Reduction of 12 with LiA1H4 in THF gave a mixture (approximately 1:7) of 13 and the  $\beta$ -epimer 13'; TLC (hexane - EtOAc, 2:1; Rf 0.42 (12), 0.29 (13) and 0.53 (13')]; the latter was not convertible to the desired product 16. Thus, deprotection of 13 with 90% trifluoroacetic acid (room temp., 12 minutes) to give the tetraol 14, followed by selective acetylation (Ac<sub>2</sub>O, pyridine, 30°C, overnight) gave the diacetate 15 (68% overall yield); mp 171~172°C, PMR (CDCl<sub>3</sub>)  $\delta$  0.93, 1.00, 1.21 and 1.29 (each 3H, s, Me  $\times$  4), 2.02 and 2.18 (each 3H, s, OAc  $\times$  2), 2.34 and 2.74 (each 1H, AB-q, J=20 Hz, CH<sub>2</sub>-9), 4.97 (1H, d, J<sub>6,7</sub> = 7 Hz, H-7), 5.08 (1H, d,  $J_{2,3} = 8$  Hz, H-3). Elimination of the hydroxyl groups of 15 with methanesulfonyl chloride (3 eq.) and 4-dimethylaminopyridine (1 eq.) in pyridine (50~80°C, 2 days) afforded the unsaturated ketone 16 (46% yield); IR (CHCl<sub>3</sub>) 1735, 1703, 1630 and 875 cm<sup>-1</sup>, UV  $\lambda_{\text{Max}}^{\text{MeOH}}$ 248 nm (ε 12,900), PMR (CDCl<sub>3</sub>) δ 1.00, 1.09 and 1.47 (each 3H, s, Me  $\times$  3), 1.6 (2H, m, CH<sub>2</sub>-5),

2.13 (6H, s, OAc  $\times$  2), 2.38 (1H, dd, J<sub>2,8</sub> = 8 Hz, J<sub>2,6</sub> = 12 Hz, H-2), 3.05 (1H, m, H-6), 5.17 (1H, s, H-9), 5.27 (1H, d, H-3), 5.63 (1H, d, J<sub>6,7</sub> = 7 Hz, H-7), 5.95 and 6.18 (each 1H, s, C=CH<sub>2</sub>).

Coriolin  $B^{2}$ , which is known to be a biogenetic analogue of coriolin (1), was treated with LiAlH4 to give the hexahydrocoriolin 17; mp 188.5~ 189.5°C (lit.<sup>2)</sup> mp 189°C); which was selectively oxidized<sup>2)</sup> with anhydrous CrO<sub>3</sub> in pyridine to yield the ketone 18; PMR (acetone- $d_6 + D_2O$ )  $\delta$ 0.88, 1.03, 1.18 and 1.30 (each 3H, s, Me  $\times$  4), 3.76 (1H, d,  $J_{2,3} = 9$  Hz, H-3), 3.80 (1H, d,  $J_{6,7} =$ 5.5 Hz, H-7), TLC [CHCl3 - MeOH, 8:1] Rf 0.31 compared with 0.35 for 14. Selective acetylation of 18 gave the diacetate 19; mp 186~187.5°C, PMR (CDCl<sub>3</sub>)  $\delta$  0.91, 1.02, 1.08 and 1.26 (each 3H, s, Me  $\times$  4), 1.90 (1H, dd, J<sub>2,3</sub> = 8 Hz, J<sub>2,6</sub> = 12 Hz, H-2), 2.04 and 2.18 (each 3H, s, OAc × 2), 2.32 and 2.90 (each 1H, AB-q, J=20 Hz, CH<sub>2</sub>-9), 2.83 (1H, m, H-6), 5.04 (1H, d, J<sub>6,7</sub> = 6.5 Hz, H-7), 5.17 (1H, d, H-3), TLC (benzene - acetone, 3:1) Rf 0.35 (compared with 0.40 for 15); which was converted to the unsaturated ketone in the manner described above. The spectral data were in total agreement with the aforementioned product 16, showing that 14 and 15 differed from 18 and 19 only at the C-11 position with respect to the relative configuration.

Finally, deacetylation of 16 in aq.THF with 1 N LiOH (room temp, 1 day), followed by epoxidation<sup>9)</sup> with excess of alkaline hydrogen peroxide (30% H<sub>2</sub>O<sub>2</sub>, NaHCO<sub>3</sub>, aq. THF, 0°C to room temp., 6 hours), completed the synthesis, giving  $(\pm)$ -coriolin (1) in 6% overall yield; mp  $181 \sim 183^{\circ}$ C (from EtOAc - CHCl<sub>3</sub> - hexane), PMR (CDCl<sub>3</sub>)  $\delta$  0.93, 1.09 and 1.23 (each 3H, s, Me  $\times$  3), 1.49 and 1.86 (each 1H, m, CH<sub>2</sub>-5), 2.32 (1H, dd,  $J_{2,3} = 9$  Hz,  $J_{2,6} = 12$  Hz, H-2), 2.75 (1H, m, H-6), 2.98 and 3.14 (2H, AB-q, J=6.5 Hz, exocyclic ethylene oxide), 3.56 (1H, s, H-9), 3.77 (1H, d, H-3), 4.05 (1H, d, J<sub>6,7</sub>=6 Hz, H-7); which was identical both spectroscopically (IR, PMR and mass spectra) and chromatographically (TLC detected by stain with H<sub>2</sub>SO<sub>4</sub> and by bioautography using Bacillus subtilis PCI 219) with an authentic sample of the natural product.

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mass spectral analyses.